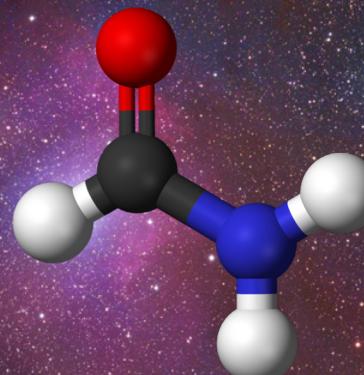
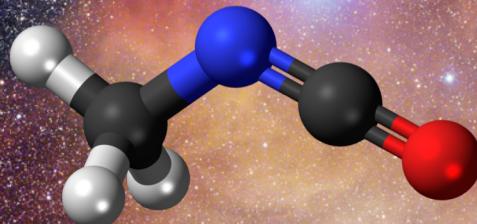




Science & Technology
Facilities Council



Chemical modelling of formamide and methyl isocyanate in star-forming regions



David Quénard
Post-Doctoral Research Assistant

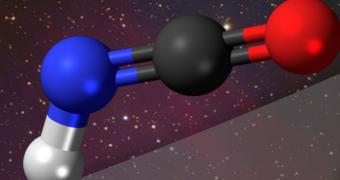
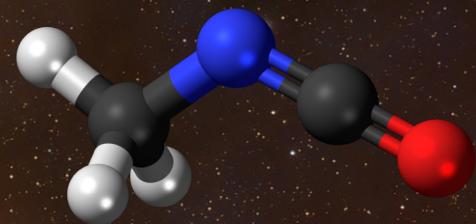
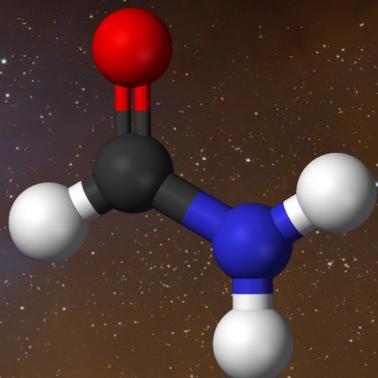
Izaskun Jiménez-Serra (QMUL), Serena Viti (UCL), Jon Holdship (UCL), Audrey Coutens (LAB)

The search for pre-biotic species

The peptide bond: CO-NH

Important bond in biochemistry
(link between two amino-acids)

Several species detected with a peptide-like bond
(e.g. NH₂CHO, CH₃NCO) or a peptide bond (HNCO)



The search for pre-biotic species

Glycine
 $(\text{NH}_2\text{CH}_2\text{COOH})$

Methylamine
 NH_2CH_3

Cyanamide
 NH_2CN

Aminoacetonitrile
 $\text{NH}_2\text{CH}_2\text{CN}$

Glycolaldehyde
 HOCH_2CHO

Methyl Formate
 HCOOCH_3

Acetic Acid
 CH_3COOH

Hydroxylamine
 NH_2OH

Acetamide
 NH_2COCH_3

Formamide
 NH_2CHO

N-Methyl Formamide
 $\text{N-CH}_3\text{NHCHO}$

Isocyanic Acid (+isomers)
 $\text{HNCO}/\text{HCNO}/\text{HO CN}$

Methyl Isocyanate
 CH_3NCO (+isomers)

→ Understand the chemistry of glycine precursors and COM-related species.

The search for pre-biotic species

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 $\text{N-CH}_3\text{NHCHO}$

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Methyl Isocyanate
 CH_3NCO (+isomers)

→ Understand the chemistry of glycine precursors and COM-related species.

Modelling of NH₂CHO, CH₃NCO, HNCO (& isomers)

UCLCHEM (Viti et al. 2004; Holdship et al. 2017)
<https://uclchem.github.io/>

Gas-phase + dust grain chemical code (364 species; 3446 reactions)

Recently proposed gas-phase/grain-surface reactions
for HNCO and CH₃NCO (+ isomers)

= grain surface

Reactions	Reference
Isocyanic Acid – HNCO/HOCN/HCNO	
Complex gas/grain network	Quan et al. (2010)
#NH + #CO → #HNCO	Fedoseev et al. (2015)
Methyl Isocyanate – CH ₃ NCO	
HNCO + CH ₃ → CH ₃ NCO + H	Halfen et al. (2015)
#CH ₃ + #OCN → #CH ₃ NCO	Belloche et al. (2017) ; Ligterink et al. (2017)
#CH ₃ + #HNCO → #CH ₃ NCO + #H	Ligterink et al. (2017)
#CH ₃ + #HNCO → #CH ₄ + #OCN	Ligterink et al. (2017)
#CH ₃ NCO + #H → #CH ₃ NH + #CO	Ligterink et al., private communication

Modelling of NH₂CHO, CH₃NCO, HNCO (& isomers)

New theoretical calculations from Majumdar et al. (2018)

Reaction		α	β	γ
HNCO + CH ₃	\rightarrow CH ₃ NCO + H	1.00×10^{-10}	0	8.04×10^3
CH ₃ NCO + H ₃ ⁺	\rightarrow CH ₃ NCOH ⁺ + H ₂	1.00×10^{-9}	-0.5	0
CH ₃ NCO + HCO ⁺	\rightarrow CH ₃ NCOH ⁺ + CO	1.09×10^{-9}	-0.5	0
CH ₃ NCO + H ⁺	\rightarrow CH ₃ NCO ⁺ + H	1.00×10^{-9}	-0.5	0
CH ₃ NCO + CO ⁺	\rightarrow CH ₃ NCO ⁺ + CO	1.00×10^{-9}	-0.5	0
CH ₃ NCO + He ⁺	\rightarrow CH ₃ NCO ⁺ + He	1.00×10^{-9}	-0.5	0
CH ₃ NCO ⁺ + e ⁻	\rightarrow CH ₃ + OCN	1.50×10^{-7}	-0.5	0
CH ₃ NCOH ⁺ + e ⁻	\rightarrow CH ₃ NCO + H	3.00×10^{-7}	-0.5	0
CH ₃ NCO + CRP	\rightarrow CH ₃ + OCN	4.00×10^3	0	0
CH ₃ NCO + Photon	\rightarrow CH ₃ + OCN	5.00×10^{-10}	0.0	0
HCN + s-CO	\rightarrow s-HCN...CO	1	0	0
s-HCN...CO + s-H	\rightarrow s-H ₂ CNCO	1	0	2.40×10^3
s-H ₂ CNCO + s-H	\rightarrow s-CH ₃ NCO	1	0	0
s-CH ₃ + s-HNCO	\rightarrow s-CH ₃ NCO	1	0	8.04×10^3
s-CH ₃ + s-OCN	\rightarrow s-CH ₃ NCO	1	0	0
s-CH ₃ + s-OCN ⁻	\rightarrow s-CH ₃ NCO + e ⁻	0	0	0
s-N + s-CH ₃ CO	\rightarrow s-CH ₃ NCO	1	0	0

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Modelling of NH₂CHO, CH₃NCO, HNCO (& isomers)

Recently proposed gas-phase/grain-surface reactions for NH₂CHO

Reactions	Reference
Formamide – NH ₂ CHO	
NH ₂ + H ₂ CO → NH ₂ CHO + H	Skouteris et al. (2017)
#HNCO + #H → #NH ₂ + #CO	Song & Kästner (2016)
#HNCO + #H → #H ₂ NCO	Song & Kästner (2016)
#H ₂ NCO + #H → #NH ₂ CHO	Song & Kästner (2016)
#H ₂ NCO + #H → #HNCO + #H ₂	Noble et al. (2016)
#NH ₂ + #HCO → #NH ₂ CHO	Fedoseev et al. (2016)
#NH ₂ + #HCO → #NH ₃ + CO	Fedoseev et al. (2016)
#NH ₂ + #H ₂ CO → #NH ₂ CHO + #H	Fedoseev et al. (2016)
#NH ₂ + #H ₂ CO → #NH ₃ + #HCO	Fedoseev et al. (2016)
#H ₂ NCO + #CH ₃ → #CH ₃ CONH ₂	Belloche et al. (2017)
#NH ₂ CHO + #OH → #H ₂ NCO + #H ₂ O	Belloche et al. (2017)
#NH ₂ CHO + #CH ₂ → #CH ₃ CONH ₂	Belloche et al. (2017)

Observational constraints

COMs in the pre-stellar core L1544

O-bearing and N-bearing COMS are **more abundant at $r \sim 4000$ AU** (methanol peak position) (Jiménez-Serra et al. 2016)

Important non-detections:

Core centre

$X [NH_2CHO] < 2.4 \times 10^{-13}$
 $X [CH_3NCO] < 2.0 \times 10^{-12}$

Methanol peak

$X [NH_2CHO] < 6.7 \times 10^{-13}$
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Methanol peak

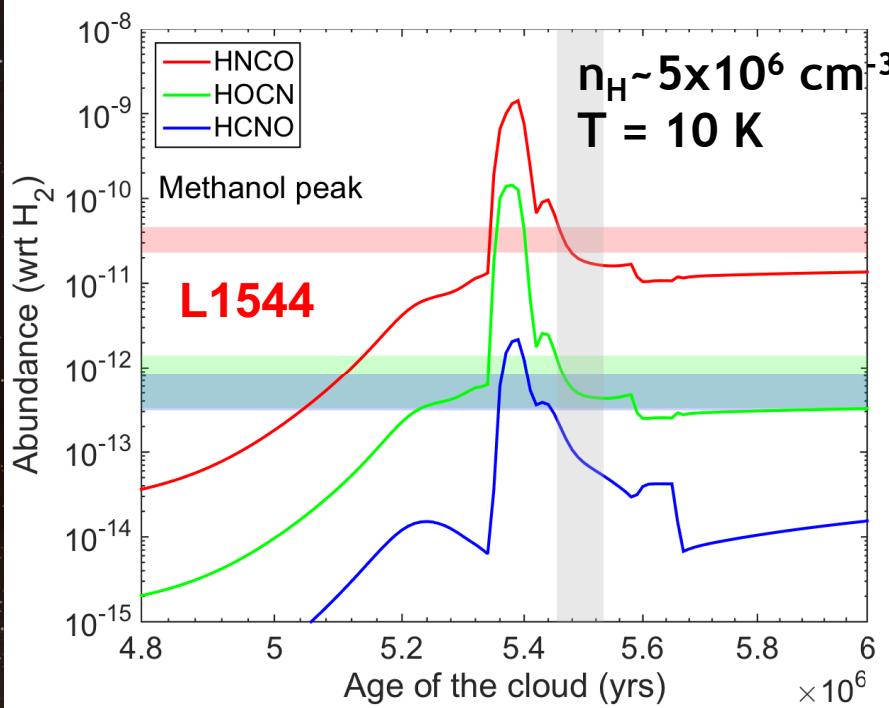
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IRAS16293-2422: hot corino and envelope

Recent detection of **CH₃NCO** toward the hot corino B!
(Martín-Doménech et al. 2017, Ligterink et al. 2017)

+

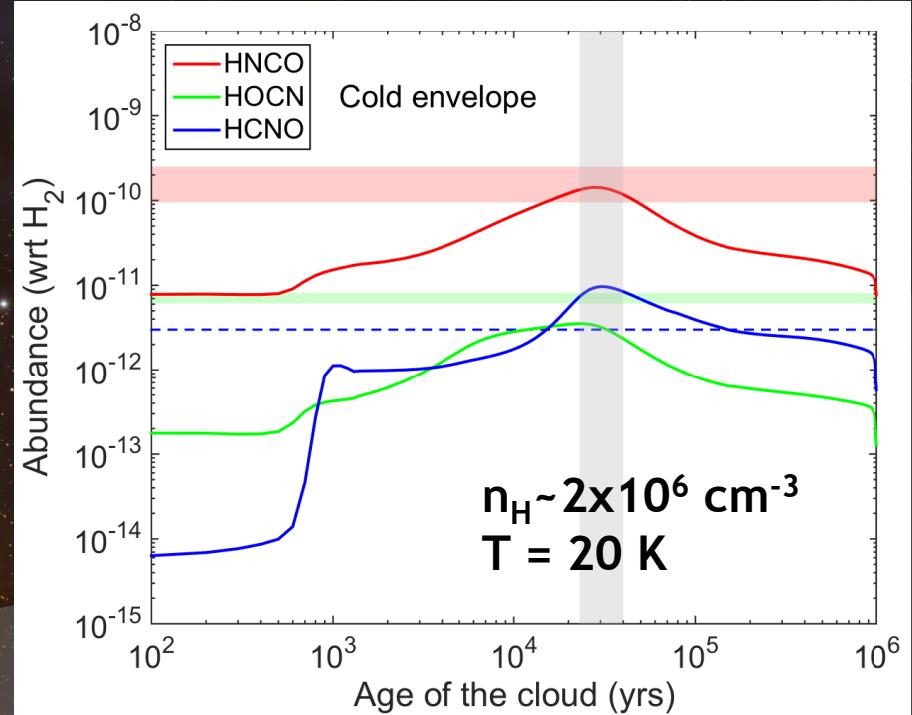
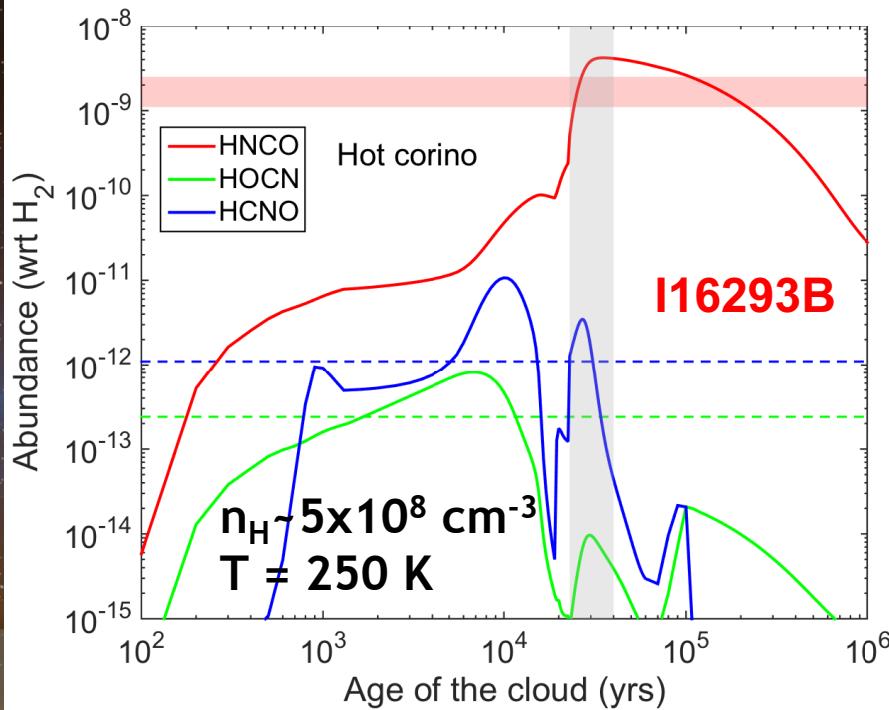
NH₂CHO observation from
Jaber et al. (2014) and Lopéz-Sepulcre et al. (2015)

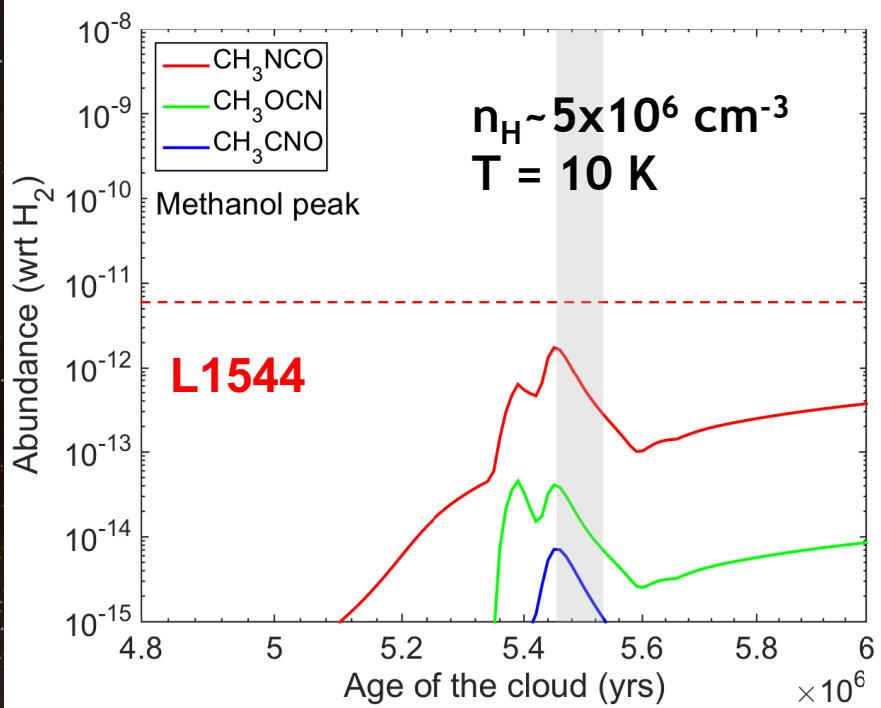


$\text{L1544} \rightarrow t \sim 5.5 \times 10^6 \text{ yr}$

$\text{IRAS16293} \rightarrow$ Same chemical
age used for both positions:
 $t \sim 3 \times 10^4 \text{ yr}$

Good agreement for HNCO in
all three regions

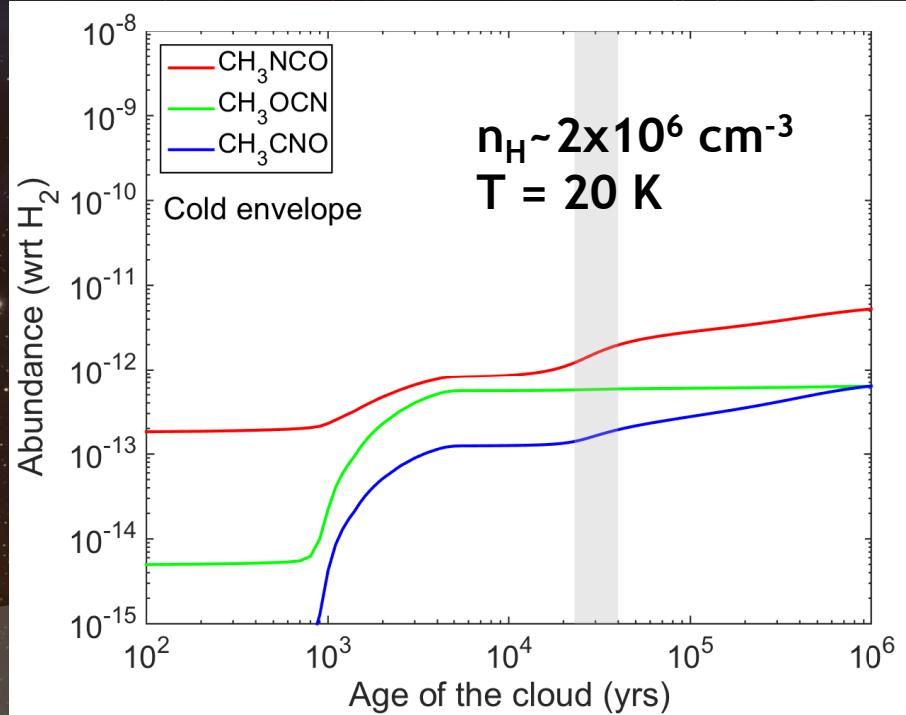
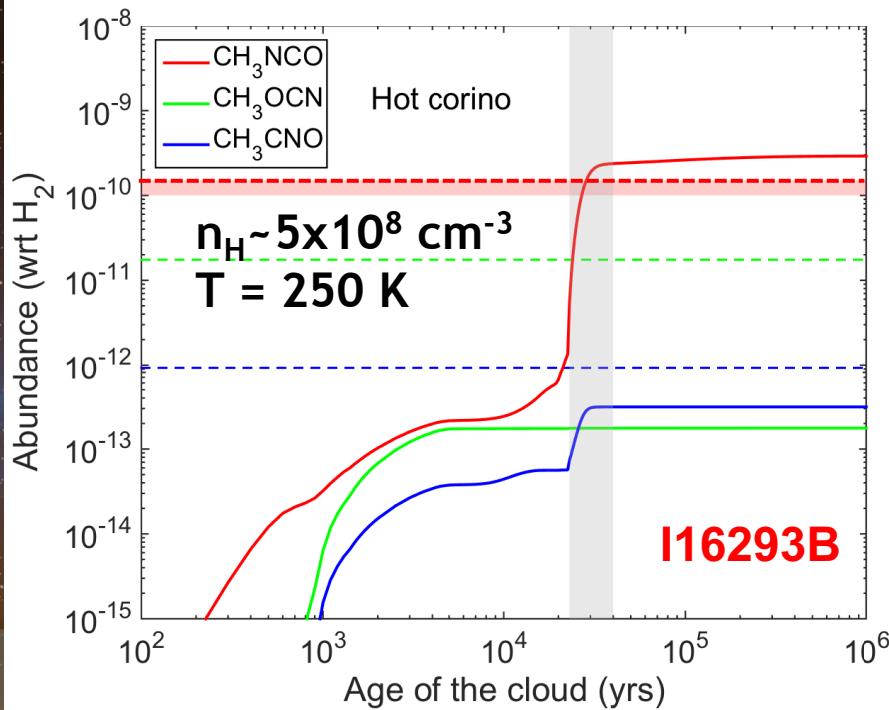


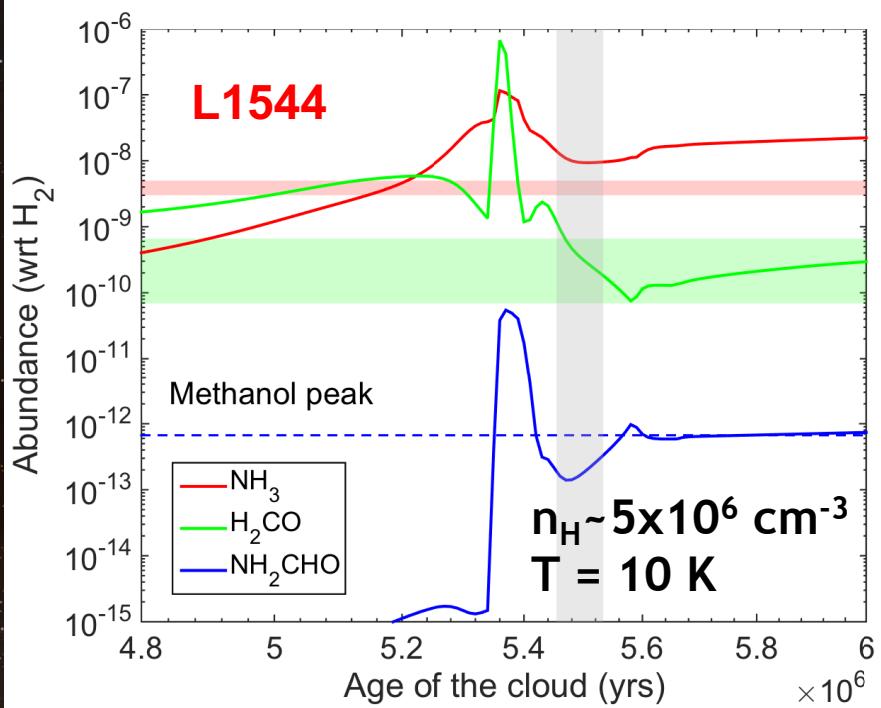


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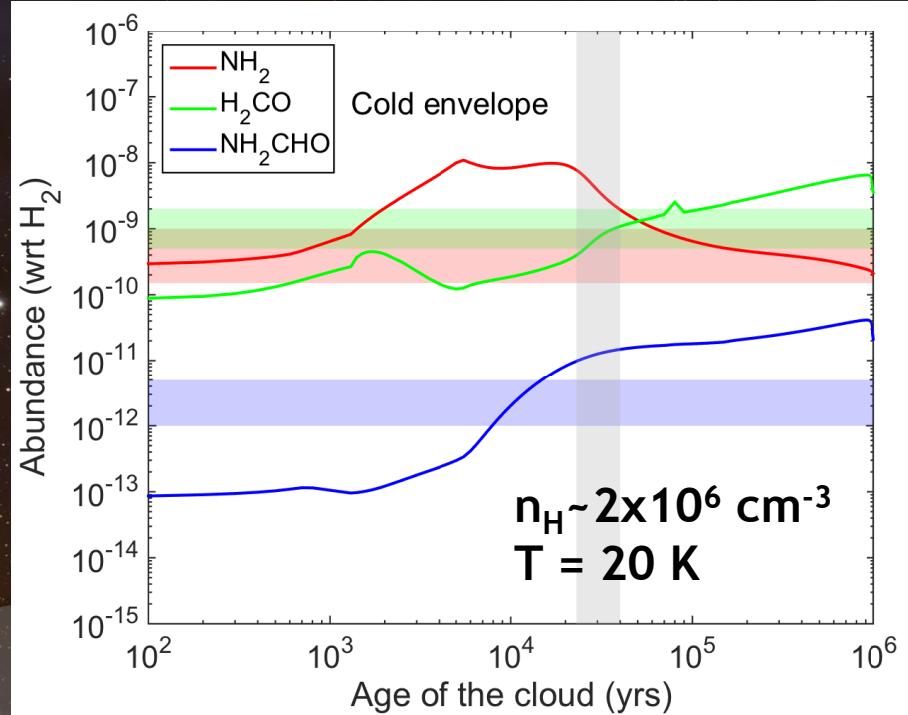
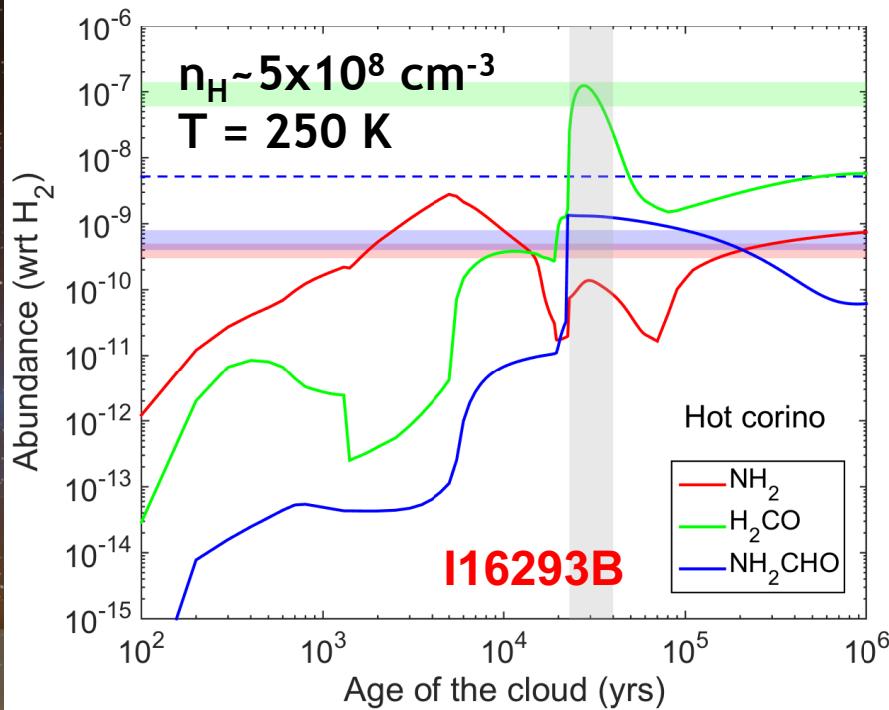




L1544 → t ~ 5.5 × 10⁶ yr

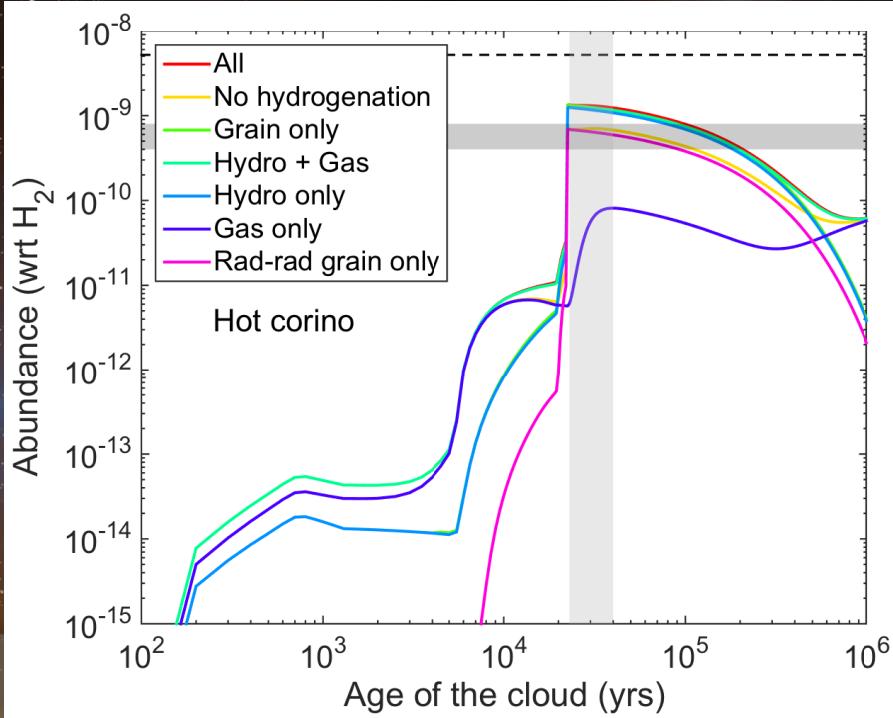
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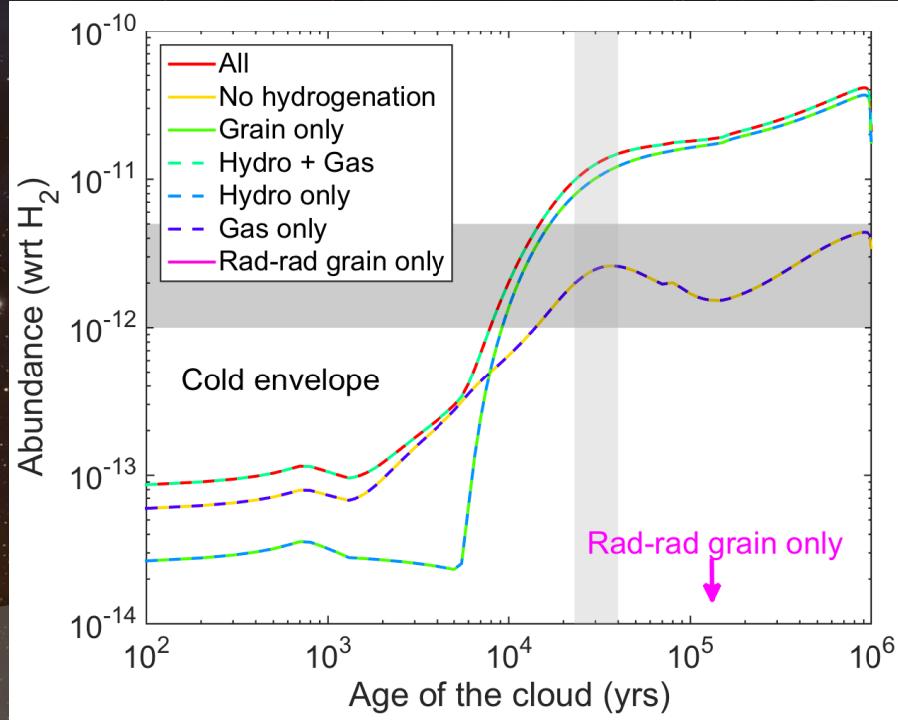
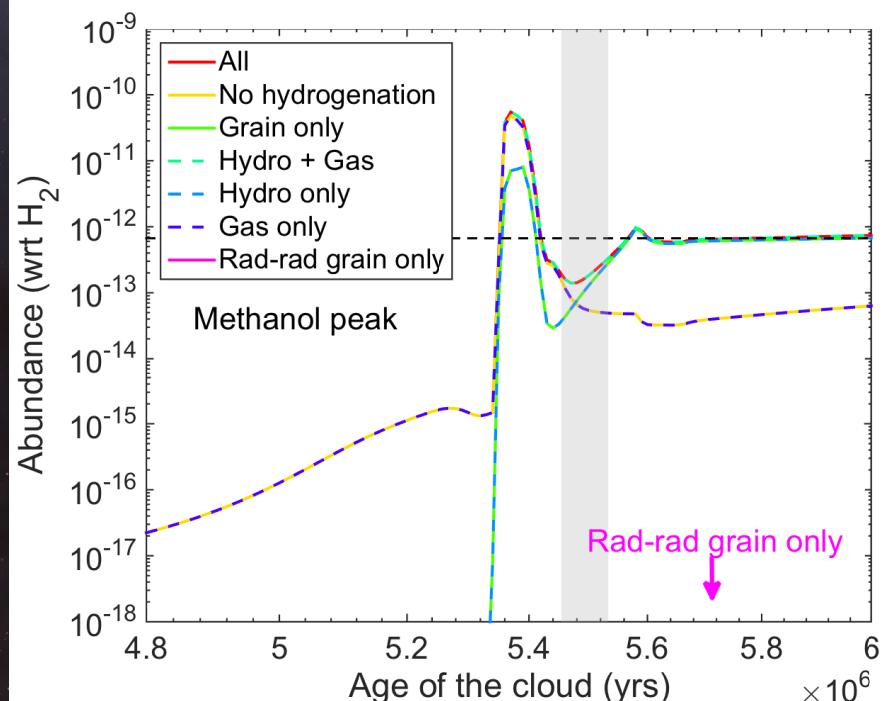
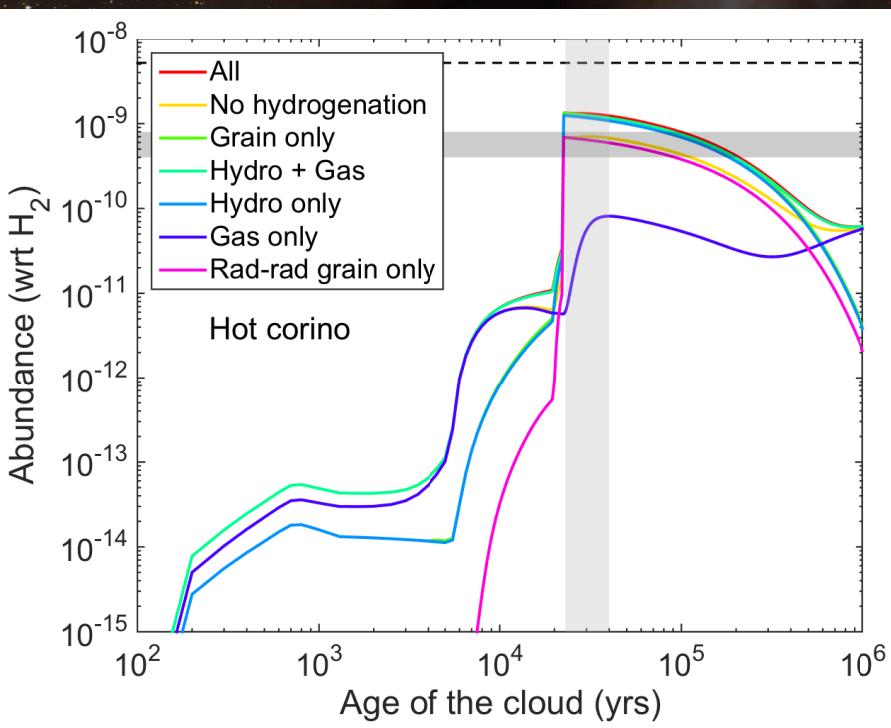
NH₂CHO chemistry

- Gas phase chemistry
- Grain surface chemistry:
 - Radical-radical reactions
 - Hydrogenation



NH₂CHO chemistry

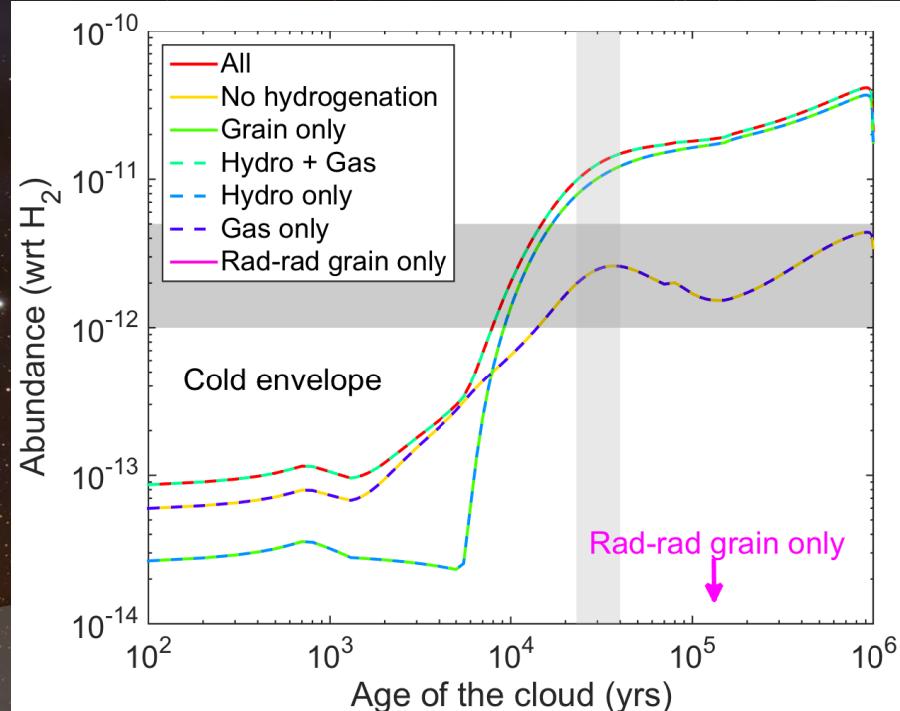
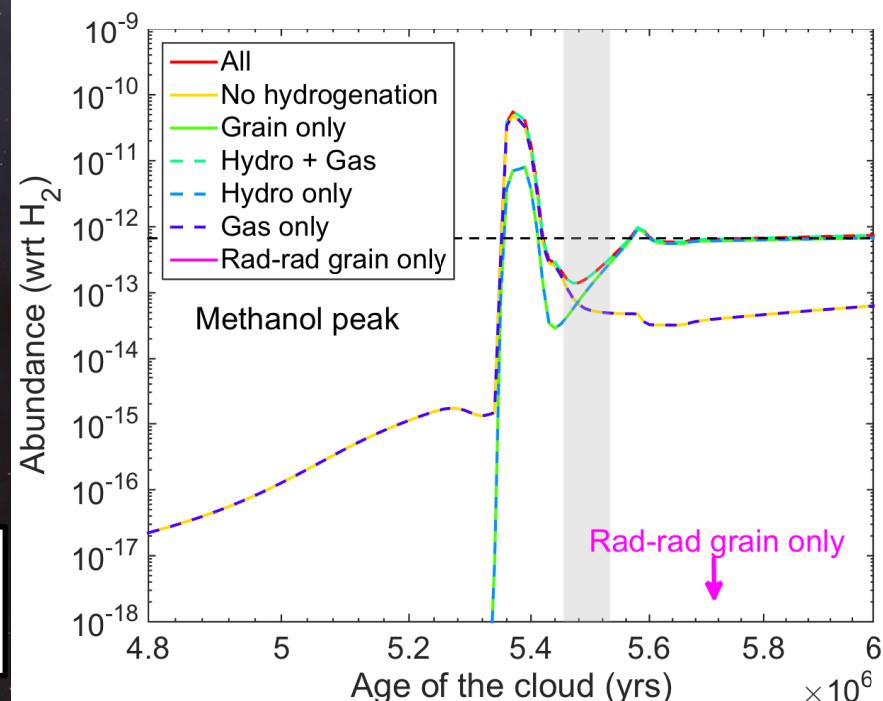
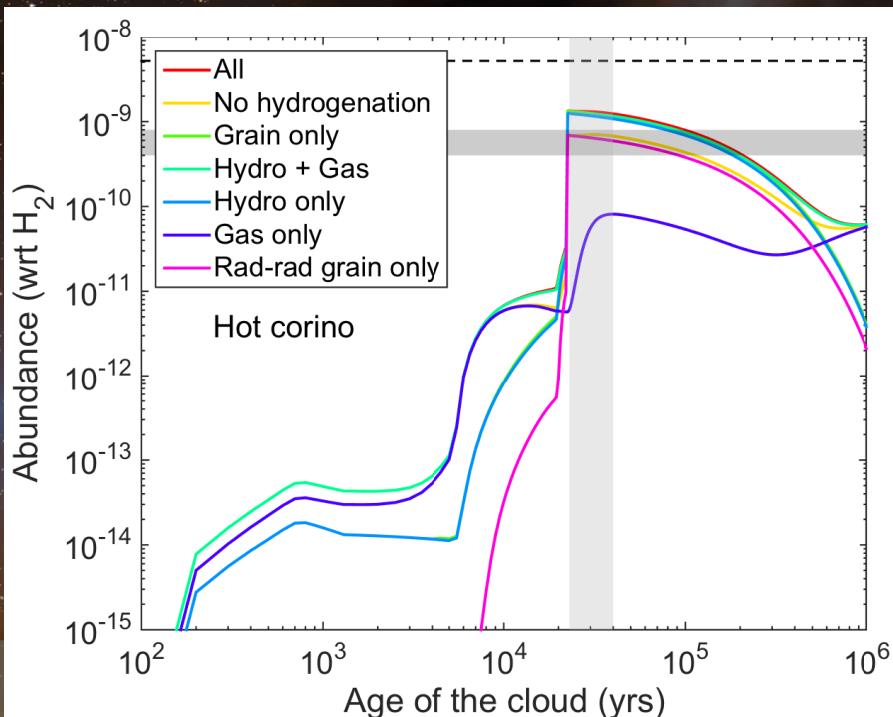
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NH₂CHO chemistry

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Modelling different physical regimes
help to constrain the chemistry !

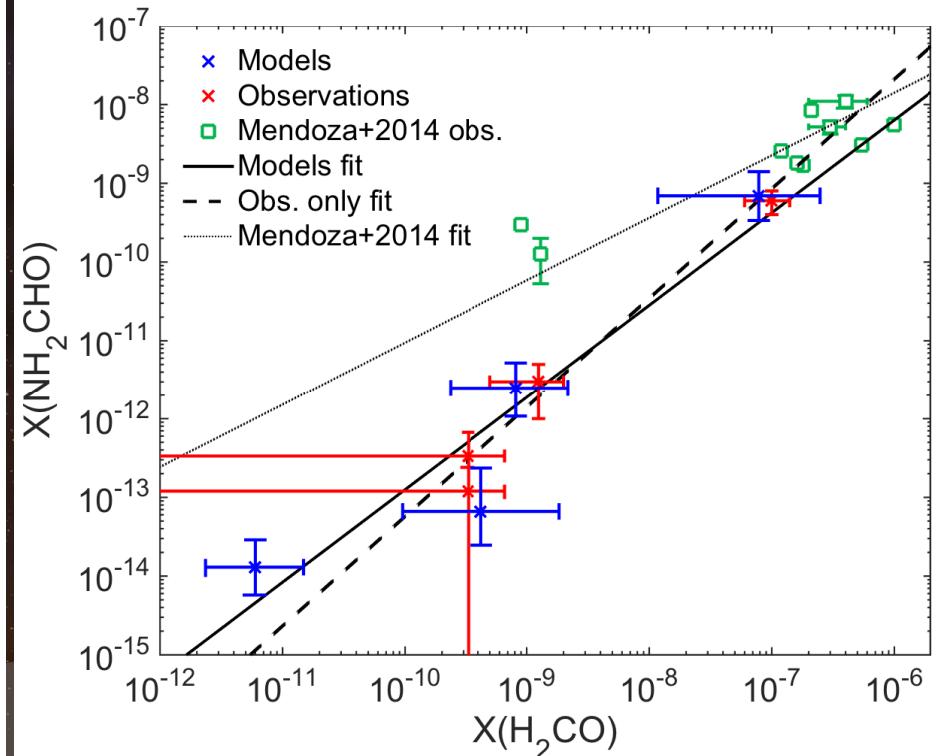
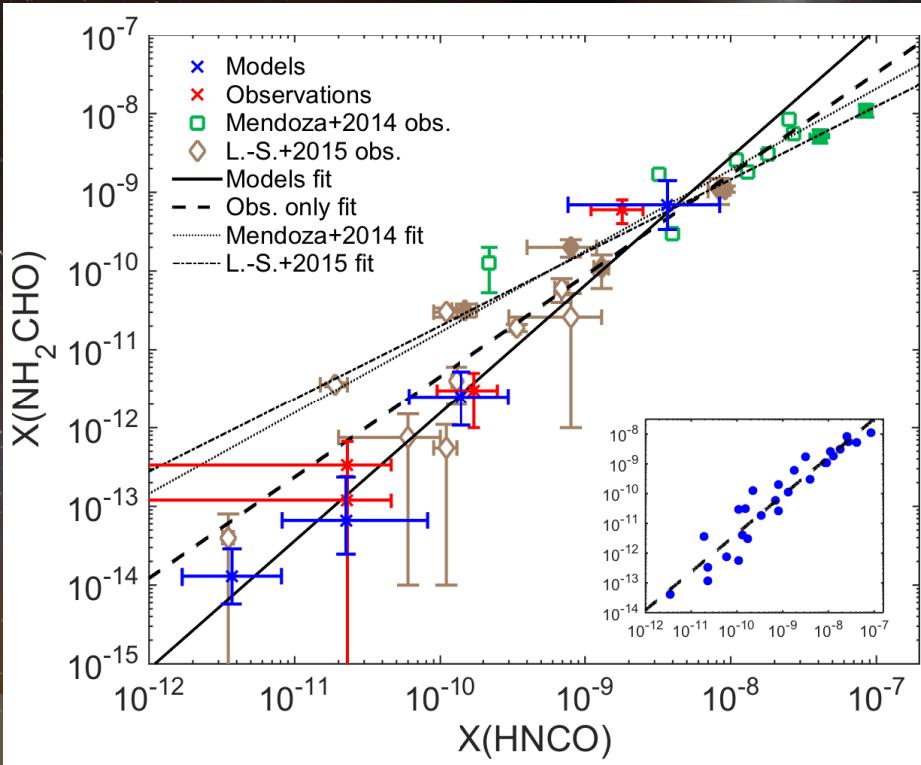


HNCO & H₂CO vs NH₂CHO

Mendoza et al. (2014) and Lopéz-Sepulcre et al. (2015):
Observational correlation → Chemical correlation between the two?

Modelling of NH₂CHO (no hydrogenation from HNCO)

→ Physical (environmental) correlation depending mainly on the temperature that triggers different chemical processes.



Conclusions

- Modelling of N-bearing COMs predicts abundances of NH_2CHO , CH_3NCO (and isomers), HNCO (and isomers) in L1544 and IRAS16293 B
 - L1544: methanol peak
 - IRAS16293 B: hot corino and cold envelope
- Both gas-phase and grain-phase chemistry are needed to explain the observed abundances of NH_2CHO
 - Hydrogenation of HNCO tends to overestimate the NH_2CHO abundance compared to radical-radical reactions
- The observed correlation between HNCO and NH_2CHO may come from an environmental correlation (temperature) rather than a chemical correlation

