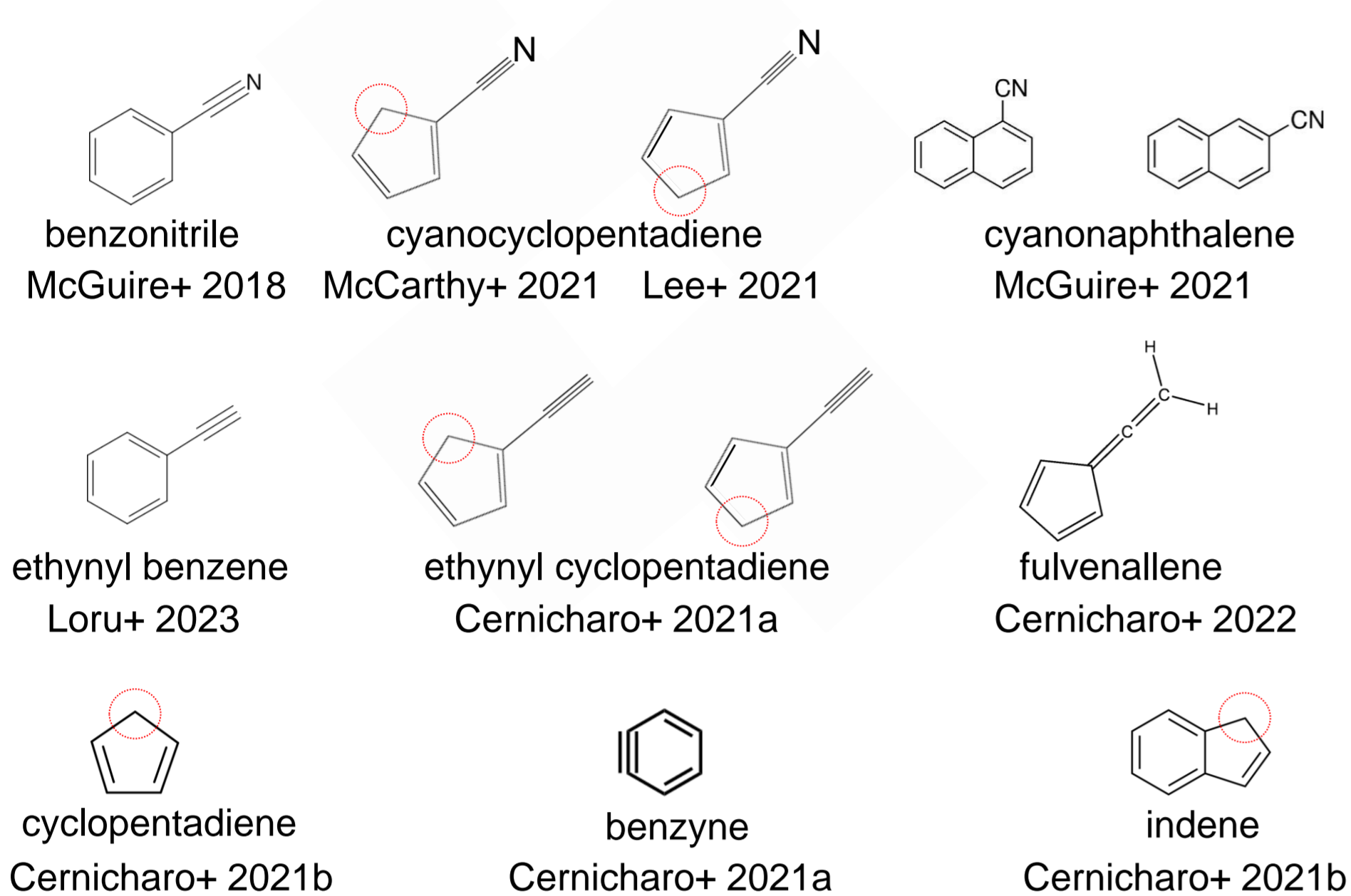




## Abstract

- This study aims to detect sulfide aromatic molecules in space.
- The rotational transitions of 2,5- and 2,4-cyclohexadien-1-thione have been measured in the 8–26 GHz region in the laboratory.
- These molecules have been produced by supersonic jet discharge using a sample of thiophenol diluted by a buffer gas of Neon.
- The rotational constants have been determined.
- The measured frequencies can now be used for astronomical detection.

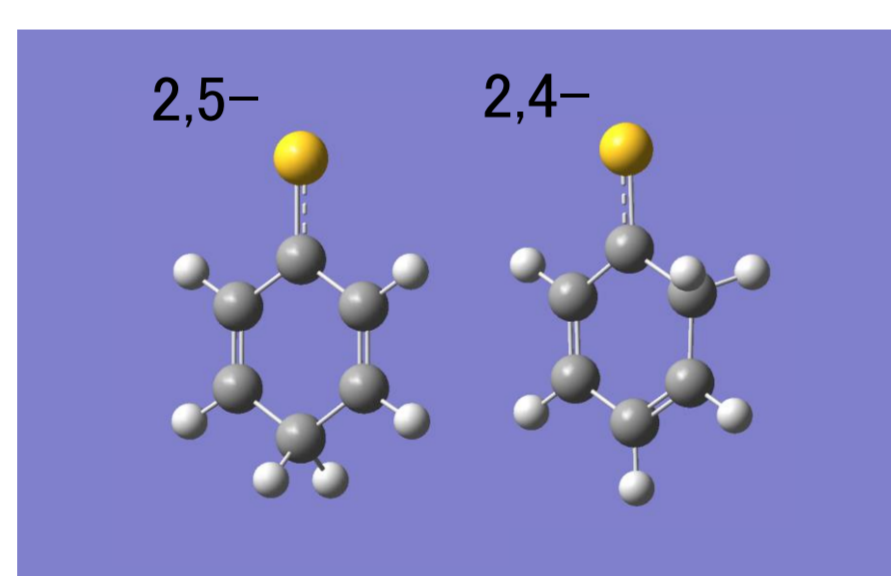
## Introduction



**Figure 1** Observed sulfide aromatic molecules in space

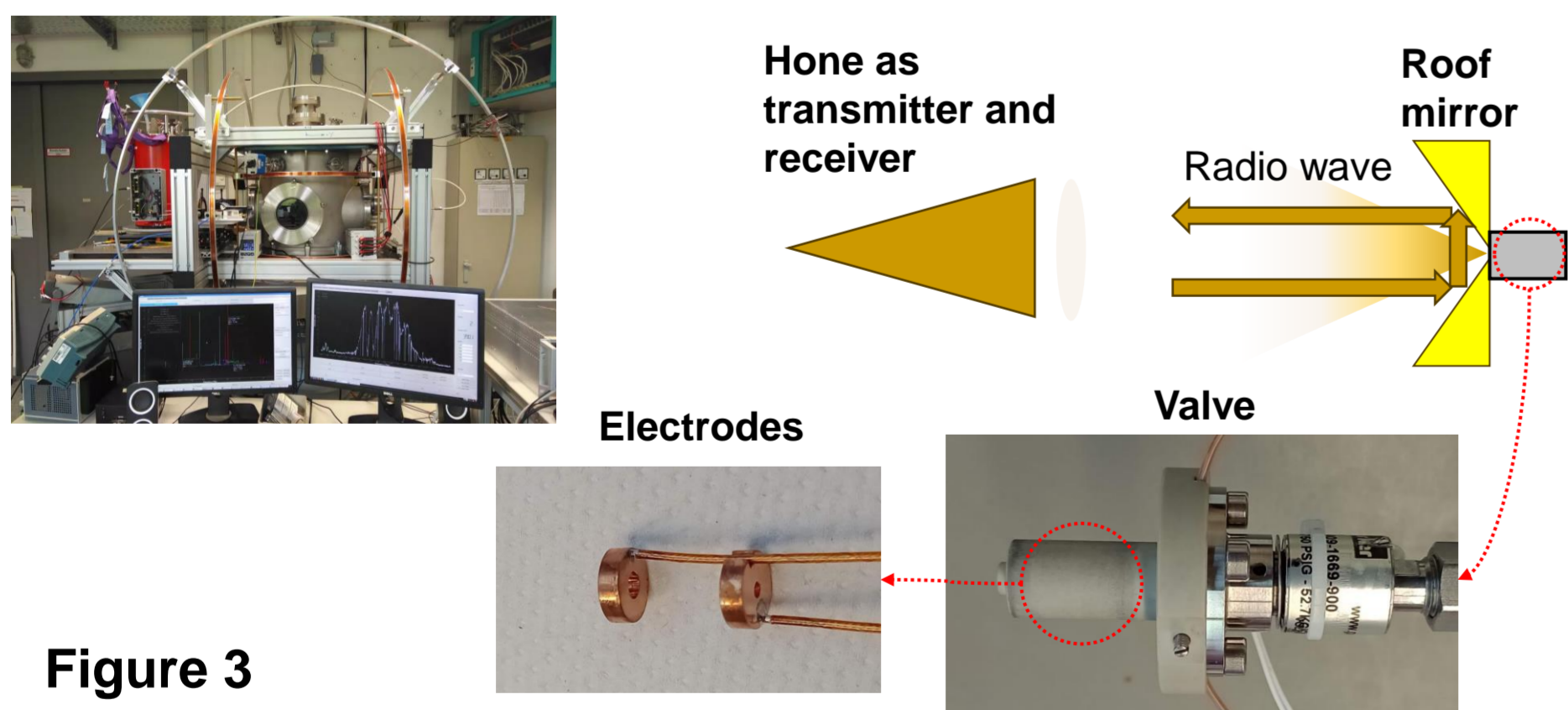
- The 325 species of interstellar molecules have been found in space (Araki).
- The most recent appearance of a new molecular group has started since the discovery of benzonitrile by McGuire et al. (2018) in the Taurus molecular cloud 1 (TMC-1). At present 12 species have been detected except for benzene (Cernicharo et al. 2001) (Figure 1). Half of them have a doubly hydrogenated carbon (-CH<sub>2</sub>-) in the ring (red circles in Figure 1).
- However, sulfide aromatic molecules have not been detected yet, although sulfide species accounts for 12% of the 325 species.
- Lack of rest frequency data of sulfide aromatic molecules by laboratory spectroscopy would be the main reason for the undiscovered.
- To find a sulfide aromatic molecule, a molecule having strong rotational lines need to be investigated.

- 2,5- and 2,4-cyclohexadien-1-thione do not have split by spin-rotation interaction, internal rotation, and hyperfine and have large dipole moments.
- The two molecules make strong rotational lines and are the easiest sulfide aromatic molecules to detect in space as a first finding.



**Figure 2** Molecular structures of 2,5- and 2,4-cyclohexadien-1-thione

## Experimental

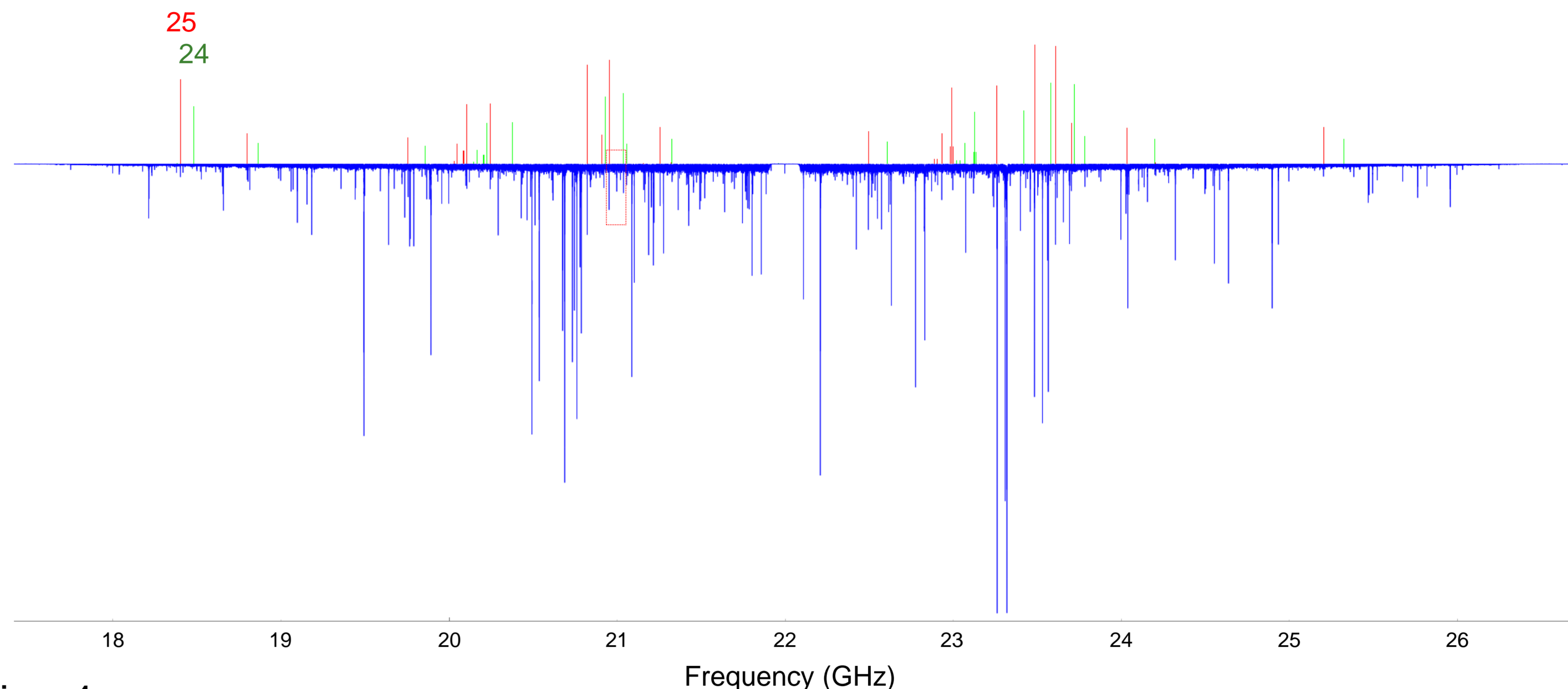


**Figure 3**

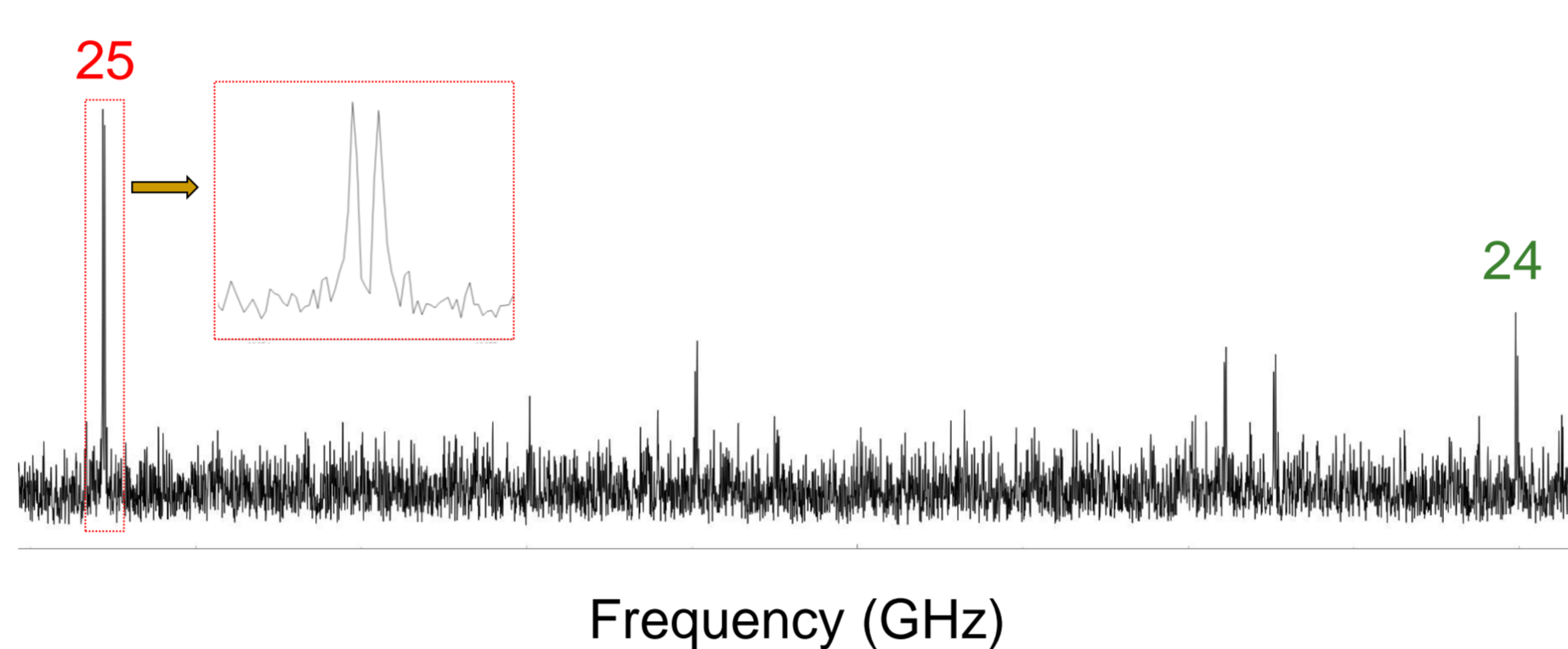
- Discharge: 10 Hz pulsed jet
- Temperature: 5 K
- Back pressure: 1.5 atom
- Voltage: 1000 V
- Current: 3–8 mA with a 50-kΩ ballast register
- Sample: thiophenol C<sub>6</sub>H<sub>5</sub>S in Ne, vapor pressure

- Lines were observed using the combination of the chirp pulsed spectrometer and the pulsed discharge supersonic jet.
- The wave source of this spectrometer consists of a synthesizer (Agilent Technologies, E8257D), an arbitrary waveform generator (Keysight, M8190A), and a solid-state amplifier (Microsemi).
- The output of the synthesizer after mixing with a chirped pulse from the arbitrary waveform generator was amplified by a solid-state amplifier and derived to a horn antenna.
- This horn antenna works as both an output port and a collecting port. Radio wave was received after interaction with gas.

## Results and Discussion



**Figure 4** Upper trace: Simulated rotational transitions of 2,5-cyclohexadien-1-thione (red) and 2,4-cyclohexadien-1-thione (green) at 5 K. Lower trace: Observed rotational lines in the 18–26 GHz region. These lines include those of the sample gas, thiophenol, and the known discharge products, CCCS, CCS, H<sub>2</sub>CS, c-C<sub>3</sub>H<sub>2</sub>, and NH<sub>3</sub>. The red dotted square area is expanded in Figure 5.



**Figure 5** Expanded spectrum of the  $J_{K_a,K_c} = 8_{1,8}-7_{1,7}$  transition for 2,5- and 2,4-cyclohexadien-1-thione. Double peaks in the red square is by Doppler splitting.

**Table 1.** Measured Frequencies of Rotational Transitions for 2,5- and 2,4-cyclohexadien-1-thione in MHz

$N_{K_a,K_c} - N_{K_a,K_c}''$	2,5-cyclohexadien-1-thione		2,4-cyclohexadien-1-thione	
	$\nu_{obs}$	$\nu_{obs} - \nu_{calc}$	$\nu_{obs}$	$\nu_{obs} - \nu_{calc}$
4 1 4-3 1 3				
4 0 4-3 0 3				
4 2 3-3 2 2				
4 2 2-3 2 1				
4 1 3-3 1 2				
5 1 5-4 1 4				
5 0 5-4 0 4				
5 2 4-4 2 3				
5 3 3-4 3 2				
5 3 2-4 3 1				
5 1 4-4 1 3				
6 1 6-5 1 5				
6 0 6-5 0 5				
6 2 5-5 2 4				
6 3 3-5 3 2				
7 1 7-6 1 6	abbr.		abbr.	
7 0 7-6 0 6				
7 2 6-6 2 5				
7 3 5-6 3 4				
7 3 4-6 3 3				
7 1 6-6 1 5				
7 2 5-6 2 4				
8 1 8-7 1 7				
8 0 8-7 0 7				
8 2 7-7 2 6				
8 3 6-7 3 5				
8 3 5-7 3 4				
9 1 9-8 1 8				
8 1 7-7 1 6				
9 0 9-8 0 8				
8 2 6-7 2 5				
9 2 8-8 2 7				
9 3 7-8 3 6				
10 1 10-9 1 9				

## References

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 Cernicharo et al. 2001, ApJ, 546, L123  
 Cernicharo et al. 2021a, A&A, 649, L15  
 Cernicharo et al. 2021b, A&A, 655, L1  
 Cernicharo et al. 2022, A&A, 663, L9  
 Lee et al. 2021, ApJL, 910, L2  
 Loru et al. 2023, A&A, 677, A166  
 McCarthy et al. 2021, Nat Astron, 5, 176  
 McGuire et al. 2018, Science, 359, 202  
 McGuire et al. 2021, Science, 371, 1265

**Table 2.**

Molecular constants of 2,5- and 2,4-cyclohexadien-1-thione in MHz

	2,5-cyclohexadien-1-thione		2,4-cyclohexadien-1-thione	
	Obs. <sup>a</sup>	Calc. <sup>b</sup>	Obs. <sup>a</sup>	Calc. <sup>b</sup>
$A_0$		5356.0		5319.2
$B_0$	abbr.	1624.3	abbr.	1630.2
$C_0$		1255.6		1257.1
$\Delta_J$	0.0 <sup>c</sup>	0.000054	0.0 <sup>c</sup>	0.000058
$\Delta_K$	0.0 <sup>c</sup>	0.001002	0.0 <sup>c</sup>	0.000928
$\Delta_{J,K}$	0.0 <sup>c</sup>	0.000225	0.0 <sup>c</sup>	0.000245
$\delta_J$	0.0 <sup>c</sup>	-0.000014	0.0 <sup>c</sup>	0.000015
$\delta_K$	0.0 <sup>c</sup>	-0.000252	0.0 <sup>c</sup>	0.000178
RMS <sup>d</sup>	0.017		0.031	

<sup>a</sup> Values in parentheses denote the uncertainties (1 $\sigma$ ) and apply to the last digit of the values.

<sup>b</sup> CAM-B3LYP/cc-pVTZ using Gaussian 16W

<sup>c</sup> Fixed

<sup>d</sup> spfit

Examples of frequency estimation in a millimeter region

$$J_{K_a,K_c} = 17_{0,17}-16_{0,16}$$

2,5- 43474.442 ± 0.460 MHz  
 2,4- 43638.774 ± 0.112 MHz

**Table 3.**

The dipole moments of 2,5- and 2,4-cyclohexadien-1-thione by CAM-B3LYP/cc-pVTZ using Gaussian 16W

	2,5-	2,4-
$\mu_a$ (D)	4.73	3.88
$\mu_b$ (D)		0.52
Relative Energy (K)	0	1780

## Summary

- The 34 and 18 rotational transitions of 2,5- and 2,4-cyclohexadien-1-thione in the centimeter wave region have been measured in the laboratory.
- Molecular constraints of both the molecules have been determined.
- Rest frequencies of the molecules have been expected accurately in a millimeter region (< 1 MHz except for the higher  $K$  transitions).

## Future plan

- The millimeter region needs to be searched in the laboratory.
- We plan to investigate these molecules in an astronomical database.

## List of the observed interstellar molecules

<http://molecules-in.space/>

